Superposition Model for the Zero-field Splitting b_2^0 of Gd^{3+} Ions in α -LiIO₃ and LiNbO₃ Crystals

Wen-Chen Zhenga,b and Shao-Yi Wua,b

Department of Material Science, Sichuan University, Chengdu 610064, P. R. China International Centre for Materials Physics, Chinese Academy of Sciences, Shenyang 110016, P. R. China

Reprint requests for W. C. Z.: E-mail: zhengwenchen@netease.com

Z. Naturforsch. 57a, 749-752 (2002); received February 6, 2002

The zero-field splitting of Gd^{3+} ions in α -LiIO₃ and LiNbO₃ crystals are studied by the superposition model. The zero-field splittings b_2^0 for the trigonal Gd^{3+} centers in both crystals are reasonably explained and the defect structures of these Gd^{3+} centers are obtained. These defect structures are consistent with the expectation based on the electrostatic interaction models and agree qualitatively with the corresponding results obtained for similar trivalent paramagnetic (rare-earth and transition-metal) ions in α -LiIO₃ and LiNbO₃ crystals.

Keywords: Electron Paramagnetic Resonance (EPR); Defect Structure; Superposition Model; Gd³⁺; α-LiIO₃; LiNbO₃.

1. Introduction

Crystals of LiNbO₃ and α -LiIO₃, activated with transition metal and rare-earth ions, are of considerable interest because they can be used for optical devices (solid state lasers, holographic data storage, amplifiers, optical waveguides, etc.). Many methods have been used to study these active or impurity centers in the two crystals [1-12]. The EPR zero-field splittings b_2^0 for Gd^{3+} ions (or the trigonal Gd^{3+} centers) in α -LiIO₃ [13] and LiNbO₃ [14] crystals were reported, however no theoretical investigations for these EPR data have been made. Since the zerofield splittings b_2^0 of paramagnetic impurities are sensitive to the local geometry of these impurity ions in crystals, useful information on the defect structures of the trigonal Gd³⁺ centers in α -LiIO₃ and LiNbO₃ crystals can be obtained by studying these EPR data. So the theoretical investigations of zero-field splittings b_2^0 for Gd^{3+} ions in both crystals are of interest and intriguing. In this paper, we explain the zero-field splittings b_2^0 and study the defect structures for the trigonal Gd³⁺ centers in α-LiIO₃ and LiNbO₃ crystals with the superposition model. The results are discussed.

2. Superposition Model

The empirical superposition model [15, 16] has been successfully used to explain the zero-field splittings of S-state (4f⁷, 3d⁵)-ions in crystals. The model assumes that the zero-field splitting b_2^m is given by a sum of axially symmetry contributions of the n ligands of the MX_n cluster only, i.e. [15, 16]

$$b_2^m = \sum_i \bar{b}_2(R_0) \left(\frac{R_0}{R_i}\right)^{t_2} K_2^m(\theta_i, \phi_i), \tag{1}$$

where $K_2^m(\theta_i, \phi_i)$ are the coordination factors [15, 16]. Thus, for S-state ions in the trigonal MX₆ cluster, we have

$$b_{2}^{0} = \left(\frac{3}{2}\right) \bar{b}_{2}(R_{0}) \left[\left(\frac{R_{0}}{R_{1}}\right)^{t_{2}} \left(3\cos^{2}\theta_{1} - 1\right) + \left(\frac{R_{0}}{R_{2}}\right)^{t_{2}} \left(3\cos^{2}\theta_{2} - 1\right) \right],$$
(2)

where R_i (i=1, 2) is the metal-ligand distance and θ_i is the angle between R_i and the C_3 axis of the studied trigonal MX₆ cluster in the crystal. t_2 is the power-law exponent and $\bar{b}_2(R_0)$ is the intrinsic parameter with the reference distance R_0 . For the Gd³⁺-O²⁻-combination [17] we have $t_2 \approx 2.5 \pm 1.5$

and $\bar{b}_2(R_0) \approx -(2000 \pm 500) \cdot 10^{-4} \text{cm}^{-1}$ with $R_0 \approx 2.699 \text{ Å}$.

3. Calculation for α -LiIO₃: Gd³⁺

Similar to trigonal iron-group ions (e.g. Cr³⁺ and Fe³⁺ [10, 11, 18]) and Er³⁺ [19] in α -LiIO₃ crystals, Gd³⁺ substitutes for Li⁺ in this lattice, and the excess charge compensation is performed by two nearest Li^+ vacancy (V_{Li}) along the C_3 -axis. Since the effective charge of the cation vacancy V_{Li} is negative, the Li⁺ and O²⁻ in the vicinity of V_{Li} should be displaced from their equilibrium positions because of the electrostatic interactions between V_{Li} and these ions. The radiofrequency discrete saturation (RFDS) studies [13] are suggested that for the Gd^{3+} center in α -LiIO₃ the two next-nearest Li⁺ ions along the C_3 -axis are shifted towards the neighbouring vacancy V_{1i} by 0.57 and 0.25 Å, respectively, from the equilibrium position owing to the electrostatic attraction between Li⁺ and V_{Li} . However, the displacement ΔX away from the $V_{\rm Li}$ for the six nearest $O^{2-}\, ions$ (i.e., the ions in the planes between Gd³⁺ and V_{Li}), and hence the defect structure of the trigonal Gd³⁺ center in α -LiIO₃ were not reported. Considering the O²⁻ displacement ΔX caused by the electrostatic repulsion between O^{2-} and V_{Li} , the structural parameters R_i and θ_i for the trigonal Gd³⁺ center in α -LiIO₃ can be calculated from the structural data $R_1^0 \approx 2.13 \text{ Å}$, $R_2^0 \ (\approx 2.11 \text{ Å}), \ \theta_1^0 \ (\approx 52.05^\circ) \ \text{and} \ \theta_2^0 \ (\approx 52.90^\circ) \ \text{of the}$ host LiO₆ cluster in α -LiIO₃ [20] and the displacement ΔX . By fitting the calculated zero-field splitting b_2^0 from the superposition model to the observed value, we obtain the displacement

$$\Delta X \approx 0.098 (29) \text{ Å}.$$
 (3)

The displacement direction agrees with those obtained for the trigonal ${\rm Cr^{3+}}$ and ${\rm Fe^{3+}}$ centers in α -LiIO $_3$ [10, 18]. The comparison of b_2^0 between calculation and experiment is shown in Table 1.

4. Calculation for LiNbO₃: Gd³⁺

The structure of LiNbO₃ is made up of irregular oxygen octahedra piled along the C_3 -axis and sharing faces. The centers of the octahedra are occupied by cations in the following sequence: Nb⁵⁺, vacancy, Li⁺,

Table 1. Zero-field splittings b_2^0 (in unit of $10^{-4} \, \mathrm{cm}^{-1}$) for the trigonal Gd³⁺ centers in α -LiIO₃ and LiNbO₃ crystals.

	Calculationa	Calculation ^b	Experiment
α-LiIO ₃	141.2	141.3	141.6 [23]
LiNbO ₃	1188	1183	1185 (13) [14]

 $^{^{\}rm a}$ Calculated by considering the ligand- or Gd $^{\rm 3+}$ -displacement but neglecting the radial extension of the impurity-ligand distances

Nb⁵⁺, vacancy, Li⁺ etc. [21]. So, the impurity ions, such as rare-earth and transition-metal ions, max occupy different sites. Since the impurity ions can influence strongly the properties of LiNbO3 crystals, knowledge of the location of the impurity and the defect structure of the impurity center are of importance. Rutherford backscattering spectrometry (RBS)/channeling, xray standing wave (XSW), proton-induced x-ray emission (PIXE), extended x-ray absorption fine structure (EXAFS), electron nuclear double resonance (ENDOR) and EPR measurements [1-7, 11]were used to study the lattice locations and defect structures for many rare-earth and transition-metal ions in LiNbO₃. It is found that all these divalent, trivalent and tetravalent paramagnetic impurity ions occputy Li+ octahedral sites. So, although the lattice location of Gd³⁺ in LiNbO₃ was not reported, we can suggest reasonably that Gd3+, like the other paramagnetic ions, occupies the Li⁺ octahedral sites. In the LiNbO₃ structure, since the electrostatic repulsive forces between Li⁺ and Nb⁵⁺ pair displace the cations from the centers of symmetry of their oxygen octahedra, the positions of Li⁺ and Nb⁵⁺ are eccentric and closer to the distinctive neighbouring vacant octahedra [11, 21]. If Li⁺ is replaced by the impurity ion carrying extra charge, the impurity ion should not occupy exactly the site of Li+, but is further displaced by ΔZ away from the center of octahedron along the C_3 -axis because the electrostatic repulsive force acting on the impurity is greater. This has been confirmed by RBS/channeling, XSW, EXAFS and EPR studies for many rare-earth and transition-metal ions in LiNbO₃ crystals [1-4, 11]. For rare-earth impurity ions in LiNbO₃ the displacement of the impurity from the Li⁺ position is also strongly dependent upon the ionic radius of the impurity [1, 2]. For Gd³⁺ in LiNbO₃ the off-center displacement ΔZ was not reported. According to the RBS/

^b Calculated by considering the ligand- or Gd³⁺-displacement and the radial extension of the impurity-ligand distances

channeling measurements [1], the displacement of the trivalent rare-earth impurity ion from the Li⁺ position can be approximately regarded as a function of the ionic radius of the impurity. Thus, from the ionic radius of Gd³⁺ [22], we can estimate the displacement to be $\Delta Z \approx 0.4$ Å. The local structural data R_i and θ_I of a trigonal Gd3+ center in LiNbO3 can also be calculated from the structural parameters $R_1^0 \ (\approx 2.238 \text{ Å}), R_2^0 \ (\approx 2.068 \text{ Å}), \theta_1^0 \ (\approx 44.57^\circ)$ and $\theta_2^0 \ (\approx 69.74^\circ)$ of the LiO₆ octahedron in the host LiNbO₃ [21] and the displacement ΔZ . If we apply the parameter $\bar{b}_2(R_0) \approx -2000 \times 10^{-4} \text{cm}^{-1}$ and $t_2 \approx 1$ (which are within the errors), the zero-field splitting b_2^0 for the trigonal Gd³⁺ center in LiNbO₃ can be reasonably explained. The comparison of b_2^0 between calculation and experiment is also shown in Table 1.

5. Discussions

We have shown that the zero-field splittings b_2^0 for the trigonal Gd^{3+} centers in α -LiIO $_3$ and LiNbO $_3$ crystals can be explained by taking into account suitable ligand (O^{2-} ions) displacements for α -LiIO $_3$: Gd^{3+} and Gd^{3+} displacements for LiNbO $_3$: Gd^{3+} . These displacements are consistent with expectations based on electrostatic interaction models and results obtained for similar trivalent paramagnetic impurities in the corresponding centers of both crystals [1–4, 10, 11, 18]. So, the above displacements and the superposition model parameters $\bar{b}_2(R_0)$ and t_2 obtained in [17] can be regarded as reasonable.

It should be pointed out that the radial extension of the metal-ligand distances R_i caused by the larger

ionic radius (\approx 0.938 Å [22]) of Gd³⁺ than that (\approx 0.68 Å [22]) of the replaced Li⁺ ion for the Gd³⁺ centers in both crystals are not considered in the above studies. In the previous papers, an approximate relationship [18, 23]

$$R \approx R_{\rm H} + (r_{\rm i} - r_{\rm h})/2 \tag{4}$$

was often used to estimate the impurity-ligand distance R in the doped crystals ($R_{\rm H}$ is the metalligand distance in the host crystal, and ${\rm r}_i$ and ${\rm r}_h$ are the ionic radius of the impurity and the replaced host ion, respectively). From this relationship, for α -LiIO₃:Gd³⁺, we obtain $R_1^0 \approx 2.259$ Å and $R_2^0 \approx 2.239$ Å by considering the above radial extension. Thus, by fitting the calculated zero-field splitting b_2^0 to the observed value, we obtain

$$\Delta X \approx 0.106 (35) \text{ Å.} \tag{5}$$

The results is very close to the above value obtained by neglecting the radial extension of the distances R_i . Calculated and experimental values of b_2^0 are given in Table 1.

For LiNbO₃: Gd³, we have $R_1^0 \approx 2.367$ Å and $R_2^0 \approx 2.197$ Å from (4). By slightly changing the power-law exponent t_2 from 1 to 1.2 (the latter is within the error of t_2), the calculated splitting b_2^0 agrees well with the observed value (see Table 1). Obviously, the consideration of the radial extensions of the metalligand distances does not change the above conclusions, and so these conclusions are suitable and rational.

- [1] J. Garcia Sole, L. E. Bausa, D. Jaque, E. Montoya, H. Murrieta, and F. Jaque, Spectrochim. Acta A54, 1571 (1998).
- [2] A. Lorenzo, H. Jaffrezic, B. Roux, G. Boulon, and J. Garcia Sole, Appl. Phys. Lett. **67**, 3735 (1995).
- [3] Th. Gog, M. Griebenow, and G. Materlik, Phys. Lett. A181, 417 (1993).
- [4] C. Prieto and C. Zaldo, Solid State Commun. 83, 819 (1992).
- [5] L. Rébouta, M. F. da Silva, J. C. Soares, M. Hage-Ali, J. P. Stoquert, P. Siffert, J. A. Sanz-Garcia, E. Diequez, and F. Agullo-Lopez, Europhys. Lett. 14, 557 (1991).
- [6] C. Prieto, C. Zaldo, P. Fessler, H. Dexpert, J. A. Sanz-Garcia, and E. D. Diequez, Phys. Rev. B43, 2594 (1991).
- [7] G. Córradi, H. Sothe, J.-M. Spaeth, and K. Polgar, Ferroelectrics 125, 295 (1992).

- [8] A. A. Mirzakhanyan and A. K. Petrosyan, Sov. Phys. Solid State 28, 904 (1986).
- [9] D. M. Daraseliya and A. Brauer, Phys. Status Solidi B109, 223 (1982).
- [10] W. C. Zheng and S. Y. Wu, J. Phys.: Condens. Matter **11**, 3127 (1999).
- [11] W. C. Zheng, J. Phys. Chem. Solids **56**, 61 (1995).
- [12] L. Arizmendi and J. M. Cabrera, Pys. Rev. **B31**, 7138 (1985).
- [13] D. L. Dzhaparidze, S. V. Alchyangyan, D. M. Daraseliya, and T. I. Shanadze, Phys. Status Solidi B158, k195 (1990).
- [14] B. Dischler, J. R. Herrington, A. Rauber, J. Schneider, and W. Urban, Solid State Commun. 12, 737 (1973).[15] D. L. Namer, and W. Lithan, Adv. Phys. 124, 742.
- [15] D. J. Newman and W. Urban, Adv. Phys. **124**, 743 (1975).
- [16] D. J. Newman and B. Ng, Rep. Prog. Phys. **52**, 699 (1989).
- [17] W. C. Zheng and S. Y. Wu, Physica **B304**, 137 (2001).

- [18] Z. M. LI and W. L. Shuen, J. Phys. Chem. Solids 57, 1673 (1996).
 [19] D. L. Dzhaparidze, S. V. Alchyangyan, D. M. Daraseliya, and T. I. Shanadze, Sov. Phys. Solid State 31, 502 (1989).
 [20] J. L. Dr. Bouer, F. Van Bolhuis, R. O. Iazecamp, and A. Vos, Acta Crystallogr. 21, 841 (1966).

- [21] S. C. Abrahams, J. M. Reddy, and J. L. Bernstein, J. Phys. Chem. Solids 27, 997 (1966).
 [22] R. C. Weast, CRC Handbook of Chemistry and Physics, CRC Press, Boca Raton 1989, pF-164.
 [23] W. C. Zheng, Physica B215, 255 (1995).